

5B2. NOx Reduction Technology

Technology overview

1. Background

Nitrogen oxide emissions are regulated, with acceptable concentration levels set according to the type of fuel and the size of the boiler. However, with the recent stiffening of regulations, some regions are subject to a "total amount control," which provides for a region-wide overall emissions level as is the case with sulfur oxide emissions. To comply with these regulations, flue

gas denitration equipment was commercialized in 1977, and ongoing efforts have been made to improve the durability of DeNOx catalysts as well as to reduce costs. The Selective Catalytic Reduction (SCR) process is used to decompose nitrogen oxides, mainly through the use of ammonia.

2. Technology

(1) Selective catalytic reduction process

Research and development: Mitsubishi Heavy Industries, Ltd.; Ishikawajima-Harima Heavy Industries Co., Ltd.; Babcock Hitachi K.K.

Overview

In this process, ammonia (NH₃) is blown into exhaust gas, allowing the ammonia (NH₃) to selectively react with nitrogen oxides NO_x (NO, NO₂), and decompose them into water (H₂O) and Nitrogen (N₂). In the DeNO_x reactor, since soot and dust are present in the exhaust gas, a grid- or plate-like catalyst is mainly used, as shown in Figure 1. The catalysts, as shown in Photo 1 and Photo 2, are installed in the reactor to react with the NH₃ blown into the catalyst layer from its inlet, allowing NO_x (NO, NO₂) to breakdown into water vapor (H₂O) and nitrogen (N₂). The catalyst is mainly composed of TiO₂, to which vanadium (V), tungsten (W), and the like are added as active ingredients.



The temperature at which the catalyst attains optimal performance is 350°C. At a temperature lower than this, SO₃ in the exhaust gas reacts with NH₃, producing ammonium hydrogen sulfate (NH₄HSO₄) that covers the surface of the catalyst, thereby reducing the ability to remove NO_x. At a temperature higher than 350°C, the NH₄HSO₄ decomposes, improving the removal of NO_x regardless of the SO₃ concentration. At a temperature above 400°C, NH₃ is oxidized and its volume decreases, thereby reducing its ability to remove NO_x. The

process is also designed to limit NH₃ leaks from the reactor to 5ppm or less. If a significant quantity should leak, it will react with the SO₃ in the exhaust gas, producing NH₄HSO₄, which clogs the piping when separated out by an air pre-heater.

The NO_x removal efficiency is around 80-90% for pulverized coal-fired thermal power plants. On the other hand, measures to equally disperse and mix the NH₃ with the exhaust gas as well as to create greater uniformity of the exhaust gas flows to cope with growing boiler sizes have been developed. These include placing a current plate, called a "guide vane" at the gas inlet, or dividing the gas inlet into grids, each to be equipped with an NH₃ injection nozzle.

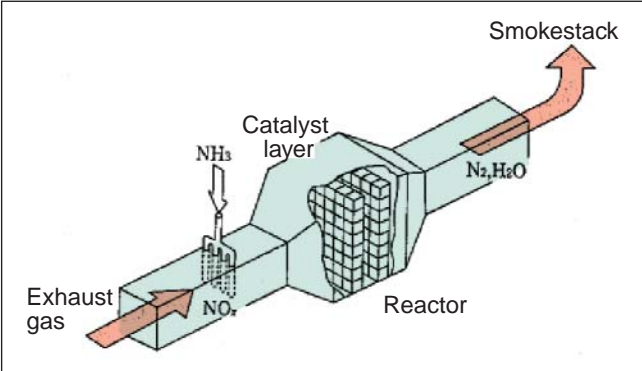


Fig. 1 Selective contact reduction method denitration process

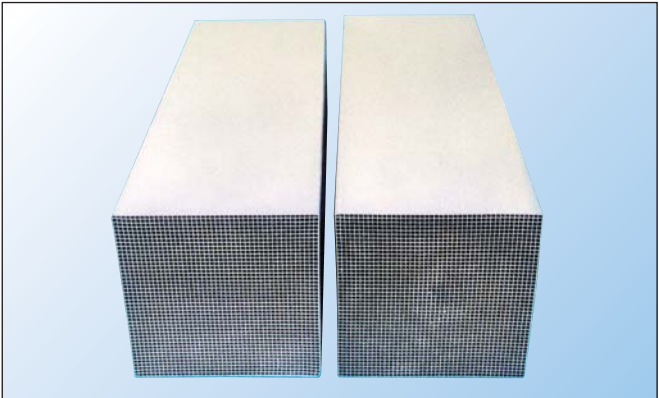


Photo 1 Grid-like catalyst

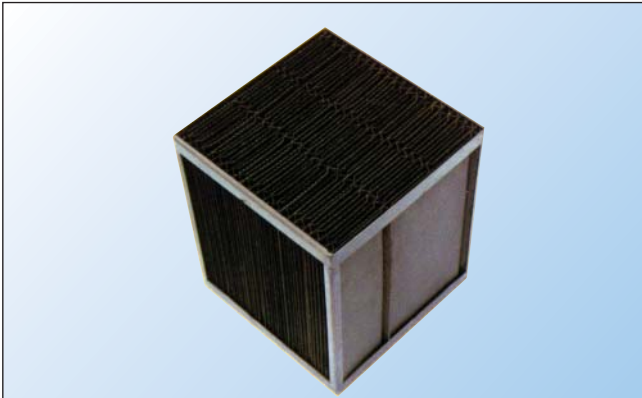


Photo 2 Plate-like catalyst

(2) Selective non-catalytic reduction (SNCR) process

Research and development: Chubu Electric Power Co., Inc.; Mitsubishi Heavy Industries, Ltd.

Project type: Voluntary

Overview

SNCR is a process to, as shown in Figure 2, blow NH_3 into the boiler section where the exhaust gas temperature is $850\text{--}950^\circ\text{C}$ and to breakdown NO_x into N_2 and H_2O without the use of a catalyst. Despite the advantages of not requiring a catalyst and its lower installation costs, the NO_x removal efficiency is as low as 40% at an NH_3/NO_x molar ratio of 1.5. Because of this, it is used in regions or equipment where there is no need for a high NO_x removal efficiency. More NH_3 is also leaked than with the selective contact reduction method, requiring measures to cope with NH_4HSO_4 precipitation in the event of high SO_3 concentrations in the exhaust gas.

This technology is mainly used at small commercial boilers and refuse incinerators. With respect to thermal power plant applications, this technology has only been installed at Chubu Electric Power's Chita thermal power plant No. 2 unit (375kw), in 1977.

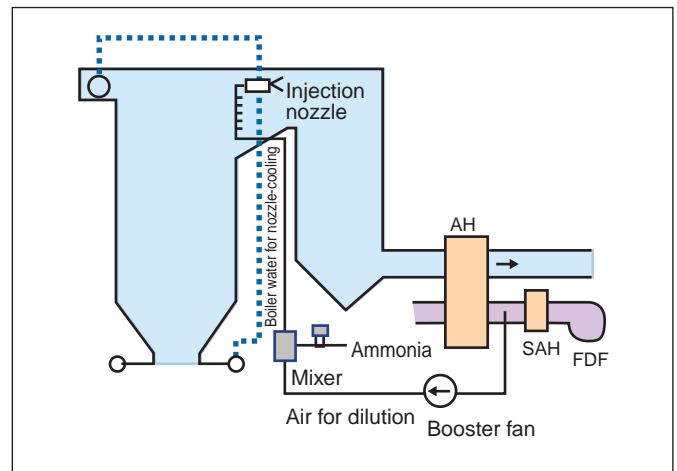


Fig. 2 Selective non-catalytic denitration NO_x removal process

(3) Radical injection method

Research and development: Japan Coal Energy Center

Project type: Subsidized coal production/utilization technology promotion

Development period: 1999-2002

Overview

In the radical injection method, as shown in Figure 3, argon plasma is injected into NH_3 , generating NH_2 plasma and other plasmas, which are then blown into the boiler to decompose NO_x into N_2 and H_2O . The target for this technology is to attain an NO_x concentration of 10ppm or less.

At present, basic research is underway at the Japan Coal Energy Center, with commercialization expected around 2010.

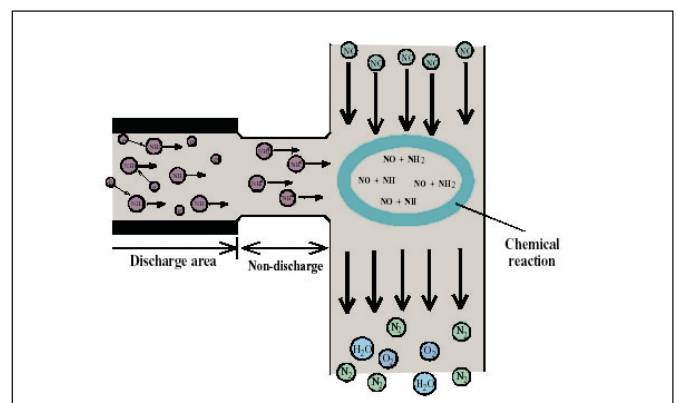


Fig. 3 Overview of radical injection method

References

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- 4) Masayuki Hirano, "New Flue Gas Denitrator Technologies for Large Boilers/Gas Turbines," Thermal/Electronic Power Generation, Vol. 50, No. 8, 1999.